

C₂-symmetric bisamidines: Chiral Brønsted bases catalysing the Diels-Alder reaction of anthrones

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Full Research Paper

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Abstract

C₂-symmetric bisamidines **8** have been tested as chiral Brønsted bases in the Diels-Alder reaction of anthrones and *N*-substituted maleimides. High yields of cycloadducts and significant asymmetric inductions up to 76% *ee* are accessible. The proposed mechanism involves proton transfer between anthrone and bisamidine, association of the resulting ions and finally a cycloaddition step stereoselectively controlled by the chiral ion pair.

Introduction

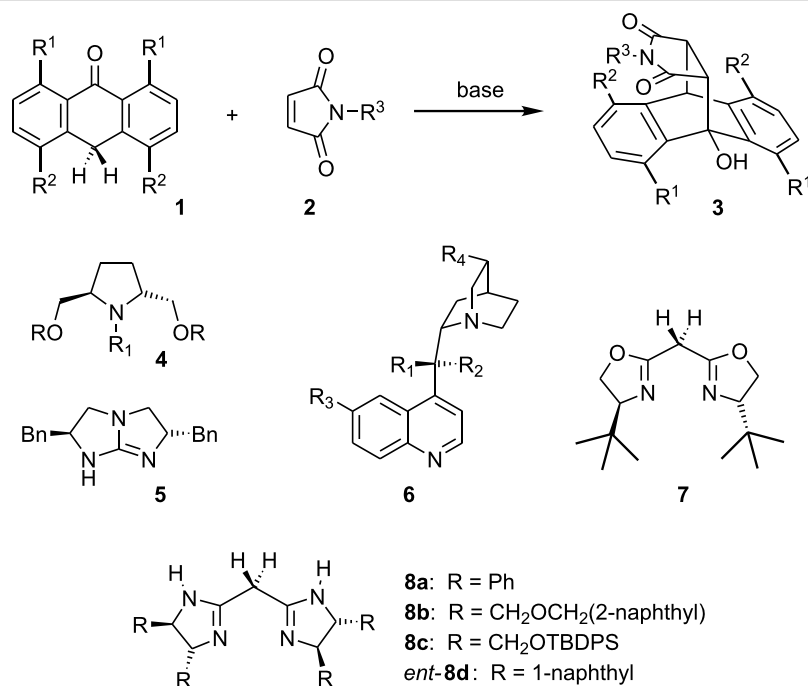
The cycloadditions of anthrones **1** and *N*-substituted maleimides **2** are prominent examples of asymmetric catalysis exerted by chiral Brønsted bases. Moderate to excellent stereoselectivities of products **3** have been reported using pyrrolidines **4** [1,2], cyclic guanidine **5** [3], or cinchona alkaloids **6** [4] as catalysts. Recently, we could promote this type of cycloaddition by metal-free bisoxazolines **7** in up to 70% *ee*, in spite of their limited Brønsted-basicity [5] (Scheme 1).

Our study was motivated by the structural similarity of bisoxazolines **7** and bisamidines **8**. Bisamidines **8**, readily accessible from malonodinitrile in two steps, prefer the conjugated tautomeric form (enamine-imine) in the monoprotinated state, which is characterised by an almost planar structure [6] (Scheme 2).

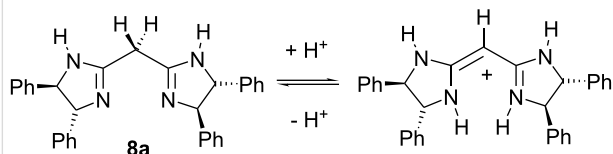
The aqueous p*K*_a of **8**·H⁺ is approximately 11, sufficient to allow deprotonation of anthrones **1** (p*K*_a around 10, [7,8]) by bisamidines to a significant extent. Here we report on the use of neutral bisamidines **8** as asymmetric Brønsted base catalysts in the cycloaddition of anthrones **1** and maleimides **2**.

Results and Discussion

Analogous to the synthesis of compound **8a** [6], the other bisamidines were prepared as hydrochlorides in 60–79% yield from the corresponding chiral diamines **9** and bisimidate **10** in refluxing ethanol. Simple extraction in the presence of Na₂CO₃ afforded the neutral bases **8b–c** and *ent*-**8d** in almost quantitative yield. The *S,S* configured diamines **9b** and **9c** were prepared from L-(+)-tartaric acid (*R,R*) via the vicinal diazide using Saalfrank's procedure [9]. **9d** was purchased as the



Scheme 1: Diels-Alder reaction of anthrones **1** and maleimides **2** catalyzed by chiral Brønsted bases **4–8**.



Scheme 2: Protonation states and tautomerism of C₂-symmetric bisamidinium **8a** [6].

dihydrochloride salt and then deprotonated by aqueous sodium hydroxide. As an “artefact” of the sequence rule, the *S,S* configured diamine **9d** leads to bisamidinium *ent-8d* (Scheme 3).

The anthrones **1b** (R¹: H; R²: Cl) and **1c** (R¹: Cl; R²: H) resulted from regioselective reductions of 1,8-dichloroanthraquinone [10,11]. Aliphatic side chains of compounds **2** could be introduced by a Mitsunobu alkylation of maleimide [12]. Alternatively, substituted maleimides were prepared by reaction of maleic anhydride with the corresponding amines followed by ring closure [13,14].

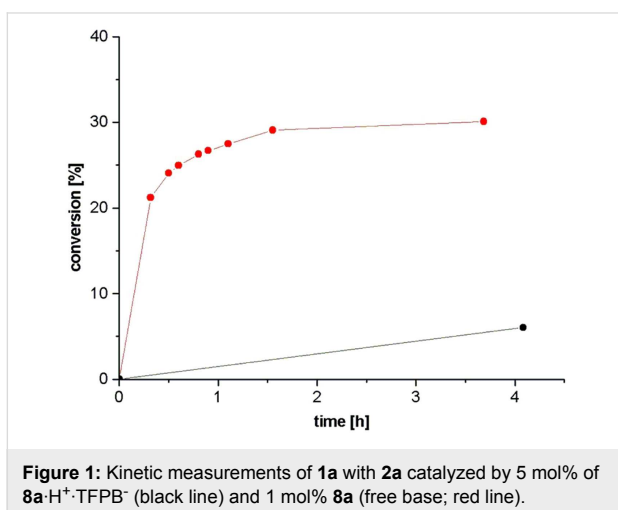
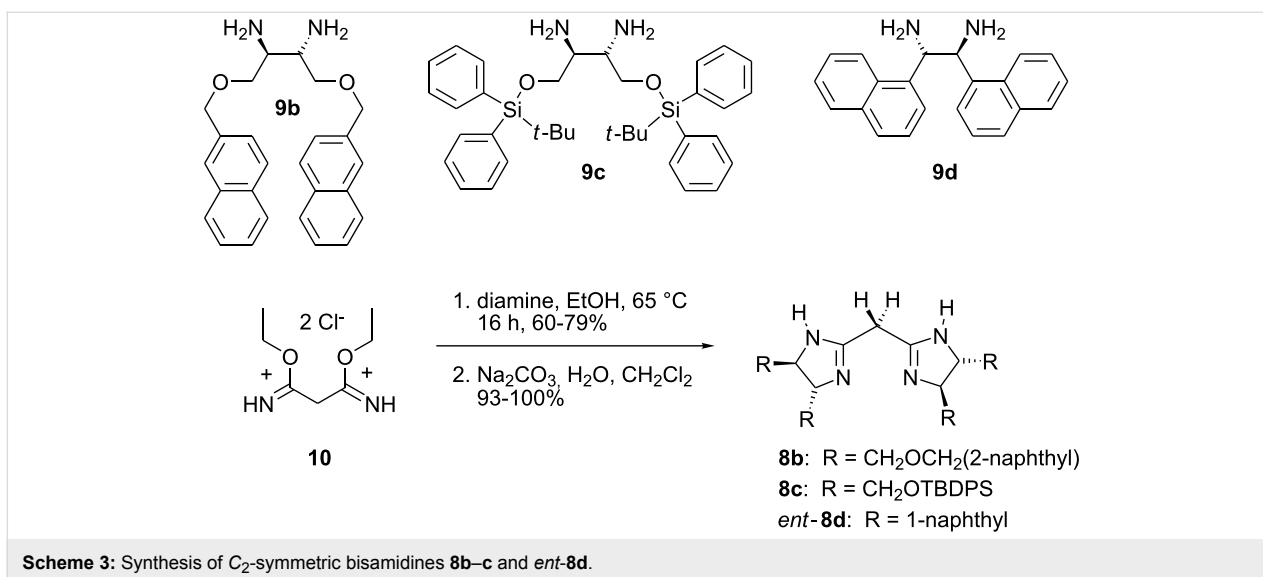
Cycloaddition kinetics of **1a** and **2a** was examined first by ¹H NMR in CD₂Cl₂ at room temperature. In the absence of catalyst, no product could be observed after 4 days. 5 mol% of the bisamidinium salt **8a**·H⁺ with tetrakis(3,5-bis(trifluoromethyl)phenyl)borate (TFPB[−]) as weakly coordinating anion resulted in 7% yield of **3a** after 4 h. In contrast, only

1 mol% of the free Brønsted base **8a** led to a high rate increase in the first 30 min. After 90 min no further conversion was observed indicating product inhibition (Figure 1). Accordingly, the reaction runs best in the base-catalyzed mode. Compared to the bisoxazolines **7**, bisamidines **8** as stronger Brønsted bases induced much higher rates in all subsequent experiments.

In the next series of experiments, bisamidines **8a–c** and *ent-8d* were compared as catalysts of the cycloaddition forming **3a** from *N*-phenylmaleimide (**2a**) and anthrone (**1a**). Using 0.25 equiv of catalyst at room temperature, isolated yields between 71% and 86% were obtained after 30 min. The best enantioselectivity, albeit low, was induced by amidine **8c** (24% *ee*). As expected, in the presence of catalyst *ent-8d* product *ent-3a* was formed preferentially (Table 1).

In a solvent screening using 10 mol% of TBDPS-protected bisamidinium **8c**, best results were obtained in dichloromethane (84% yield; 30% *ee*). Even higher yields were accessible in aromatic solvents, however, at the price of reduced stereoselectivity (Table 2).

Lowering the reaction temperature from 23 to −20 °C (**8c**, dichloromethane) retarded the cycloaddition but did not change enantioselectivities. After extended reaction times, excellent yields were still observed. Up to 39% *ee* was finally obtained at −70 °C. However, such conditions resulted in lower yields, even with increased catalyst loads and further extended reac-



tion times. Best results, 96% yield and 36% *ee* with only 10 mol% of catalyst, were found at −40 °C (Table 3).

Having identified suitable experimental conditions, we explored the scope of the bisamidine-catalyzed Diels-Alder reaction. The results are summarized in Table 4. Both electron-donating and electron-withdrawing substituents were tolerated and furnished products in good to excellent yields and with moderate values of *ee*. A remarkable increase in enantioselectivity was observed using maleimide **2i**. The steric hindrance imposed by the large 2,6-diisopropylphenyl moiety of **2i** resulted in 76% *ee* at −70 °C but also lowered reaction rates.

Only 13% yield could be obtained under such conditions. Yields rose to 65% at room temperature (51% *ee*; entries 11 and 12). With other sterically hindered dienophiles such as *N*-tert-

Table 1: First evaluation step of chiral bisamidine catalysts.

entry ^a	catalyst	yield [%] ^b	<i>ee</i> [%] ^c
1	8a	86	11
2	8b	78	17
3	8c	85	24
4	ent-8d	71	−17 ^d

^aAll reactions were carried out using 0.1 mmol maleimide **2a**, 1.1 equiv anthrone (**1a**) and 0.25 equiv of catalyst in 1 mL abs. dichloromethane at room temperature for 30 minutes. ^bIsolated yield after column chromatography. ^cThe enantiomeric excess was determined by HPLC using a Chiralpak IA column. ^dA negative *ee* stands for an excess of **ent-3a**.

Table 2: Influence of the solvent on the bisamidine catalyzed Diels-Alder reaction.

entry ^a	solvent	yield [%] ^b	ee [%] ^c
1	dichloromethane	84	30
2	chloroform	86	18
3	benzene	98	21
4	toluene	99	16
5	α,α,α -trifluorotoluene	99	13
6	dibutyl ether	89	11

^aAll reactions were carried out using 0.1 mmol maleimide **2a**, 1.1 equiv anthrone (**1a**) and 0.1 equiv of **8c** in 1 mL abs. solvent at room temperature for 60 minutes. ^bIsolated yield after column chromatography.

^cThe enantiomeric excess was determined by HPLC using a Chiralpak IA column.

Table 3: Influence of temperature on the Diels-Alder reaction.

entry ^a	reaction temperature [°C]	reaction time [h]	yield [%] ^b	ee [%] ^c
1	23	1	84	30
2	0	24	96	29
3	-20	24	98	31
4	-40	48	96	36
5	-70	96	71	39

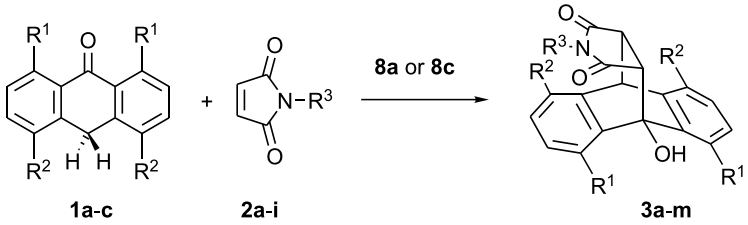
^aAll reactions were carried out using 0.1 mmol maleimide **2a**, 1.1 equiv anthrone (**1a**) and 0.1 (entry 1–4) or 0.25 equiv (entry 5) of **8c** in 1 mL abs. dichloromethane. ^bIsolated yield after column chromatography.

^cEnantiomeric excess was determined by HPLC using Chiralpak IA column.

butylmaleimide (**2c**), the level of *ee* remained low (entry 4). The halogen-substituted anthrones **1b–c** did not react with **2i** at -70 °C. At room temperature, however, **1b** and **2i** were efficiently transformed into **3m** by catalyst **8a** with 76% yield and 54% *ee*. A single recrystallisation step afforded an almost enantiopure product (96% *ee*). The *R,R* configuration of compound **3m** was determined by anomalous X-ray diffraction using a single crystal of **3m** with 96% *ee* (Figure 2).

A mechanistic rationalisation is proposed in Scheme 4. The catalyst deprotonates the anthrone in the initial step. This assumption is supported by the *pK_a* values of compounds **2a** (10, [7,8]) and **8-H⁺** (~11, [6]). Furthermore, the appearance of the yellow color of enolates (**1-H⁺**) shows significant proton transfer when bisamidine **8a** is added to anthrones **1a**, **1b**, or **1c**. A chiral contact ion pair **A** is formed and controls the stereochemical course of the Diels-Alder reaction with maleimides. In the last step, the catalyst-product-complex **B** dissociates and regenerates the unprotonated bisamidine.

Table 4: Scope of the Diels-Alder-reaction.

						
entry ^a	1 [R ¹ , R ²]	R ³	condition ^b	3	yield [%] ^c	ee [%] ^d
1	1a [H, H,]	Ph (2a)	A	3a	96	36
2	1b [H, Cl]	2a	A	3b	95	41
3	1a	iPr (2b)	B	3c	74	26
4	1a	<i>t</i> -Bu (2c)	B	3d	45	30
5	1a	Cy (2d)	B	3e	83	42
6	1c [Cl, H]	2d	B	3f	90	19
7	1a	Bn (2e)	A	3g	95	20
8	1a	CHPh ₂ (2f)	A	3h	85	26
9	1a	4-Br-(C ₆ H ₄)- (2g)	B	3i	70	13
10	1a	4-MeO-(C ₆ H ₄)- (2h)	A	3j	82	32
11	1a	2,6-iPr ₂ -(C ₆ H ₃)- (2i)	B	3k	13	76
12	1a	2i	C	3k	65	51
13	1c	2i	C	3l	77	34
14	1b	2i	C	3m	76	54 (96) ^e

^aAll reactions were carried out using 0.1 mmol maleimide, 1.1 equiv anthrone in 1 mL abs. CH₂Cl₂. ^bA = 10 mol% **8c**, -40 °C, 48 h; B = 25 mol% **8a**, -70 °C, 96 h; C = 25 mol% **8a**, r.t., 3 h. ^cIsolated yield after column chromatography. ^dThe enantiomeric excess was determined by HPLC using a Chiralpak IA column. ^eRecrystallized from 2-propanol/*n*-hexane.

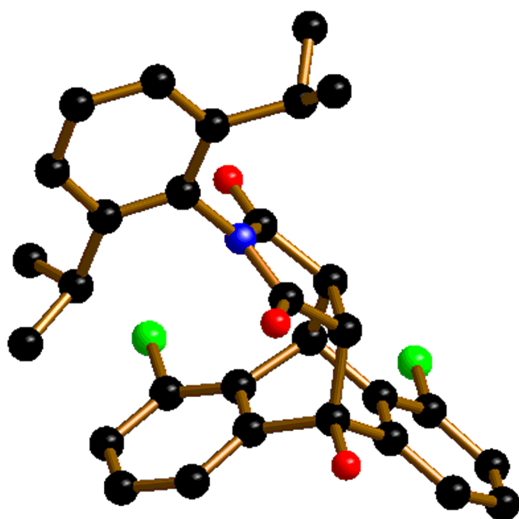


Figure 2: Molecular structure of **3m** (C: black; N: blue; O: red; Cl: green; hydrogen atoms are omitted for the sake of clarity).

Conclusion

C_2 -symmetric bisamidines were shown to be potent chiral Brønsted base catalysts for the Diels-Alder reaction of *N*-substituted maleimides and anthrones. Compared to bisoxazolines **7**, much shorter reaction times under comparable conditions were sufficient with the more basic bisamidine catalysts **8** (~50-fold [5]). The higher intrinsic reactivity of the bisamidines allowed to run the reactions at lower temperatures. In both groups of catalysts, the phenyl substituted species induced the lowest enantioselectivities. Bisamidine **8a** performed better than the

corresponding bisoxazoline. Increasing the size of substituents in catalysts **8b–d** also improved stereoselectivities, but not to high levels. This may be due to the flexible nature of the substituents present in bisamidines **8b** and **8c**. It is instructive, therefore, to compare with the bisoxazolines **7**. By far the best enantioselectivities were observed in this series with the *t*-Bu derivative (47% *ee* versus 3% for the phenyl analogue in the reaction of **1a** and **2a**). Keeping in mind that even the less selective bisamidine **8a** could induce up to 76% *ee* in favorable cases, replacing the phenyl moieties of **8a** by *t*-Bu is an attractive option for future studies on bisamidine-mediated organocatalytic transformations.

Supporting Information

Supporting Information File 1

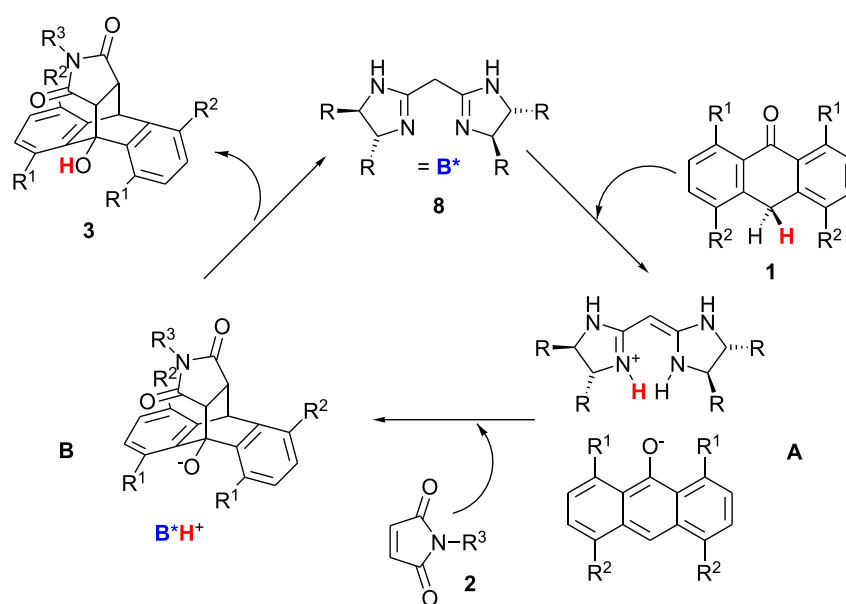
Supporting information features characterisation data and copies of ^1H - and ^{13}C -NMR spectra of anthrones **1**, maleimides **2**, Diels-Alder adducts **3**, bisamidine hydrochlorides **8b–d**· H^+Cl^- , neutral bisamidines **8b–d** and diamines **9b–c**, plus copies of chromatograms obtained with chiral columns.

[<http://www.beilstein-journals.org/bjoc/content/supplementary/1860-5397-4-28-S1.doc>]

Supporting Information File 2

X-Ray data of compound **3k**

[<http://www.beilstein-journals.org/bjoc/content/supplementary/1860-5397-4-28-S2.cif>]



Scheme 4: Proposed mechanism of the Diels-Alder reaction.

Supporting Information File 3

X-Ray data of compound **3m**

[<http://www.beilstein-journals.org/bjoc/content/supplementary/1860-5397-4-28-S3.cif>]

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